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Discussion: Buffered solutions are solutions that change pH only slightly with the addition of an acid or base. An important example of a buffered solution is human blood. Many biochemical processes take place within a narrow range of pH values, so it is important to maintain a constant blood pH. In blood, this pH range is between 7.35 and 7.45.

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Access Lakhmir Singh Solutions For Class 10 Chemistry Chapter 2. ... the yellow stain turns reddish-brown. On rinsing the shirt with plenty of water, the reddish-brown stain turns yellow again. ... 60. A first-aid manual suggests that vinegar should be used to treat wasp stings and baking soda for bee stings.

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In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

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"Organic agriculture is a production system that sustains the health of soils, ecosystems and people. It relies on ecological processes, biodiversity and cycles adapted to local conditions, rather than the use of inputs with adverse effects. Organic agriculture combines tradition, innovation and science to benefit the shared environment and promote fair relationships and a good quality of life ...

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Covalent organic frameworks (COFs) are an emerging class of crystalline porous organic materials, built up from covalently linked organic molecules. Tuning the reticular chemistry leading to these structures enables synthesis of crystalline, photoactive, porous, and highly stable materials, providing an optimal platform for photocatalysis.

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Well decomposed organic soil material with a low content of fibers (plant tissue excluding live roots). Peat.—Slightly decomposed organic soil material with a high content of original fibers. Mucky peat.—Organic soil material that is intermediate in degree of decomposition, fiber content, bulk density, and water content between muck and peat.

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The Winkler titration method of determining the amount of dissolved oxygen in water is a manual procedure that can be carried out in the field. Measured amounts of chemicals are added to the water sample to create an acid mixture that is titrated to determine the concentration of dissolved oxygen.

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The boiling points of the various water-antifreeze solutions: Have students pour 2 to 3 mL of one of the solutions to be tested in a clean test tube and add a boiling chip to prevent super-heating. Instruct them in the safe way to expose the test tube to the heat source.

[Cypermethrin | C22H19Cl2NO3 - PubChem](#)

The principle behind Immunohistochemistry (IHC) entails detection of antigen or happens in cells of a tissue section by exploiting the principle of antibodies binding specifically to antigens in biological tissues. Learn more about this in this guide.